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In the subsequent chapter they address the basics of depletion interactions, including the effects of anisotropic depletants. The next chapter covers the basics of phase behavior in colloidal dispersions, followed by extensions towards mixtures of spherical colloids with polymers or small colloids inducing depletion forces.

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Colloids are submicron particles that are ubiquitous in nature (milk, clay, blood) and industrial products (paints, drilling fluids, food). It has been known for a long time that adding salt or polymer influences the characteristics of colloidal suspensions.

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The term depletion derives from Latin meaning emptied out. The verb plere is to fill [24]. Thus a pletion force is due to accumulation of some substance between two colloids. The reversal, a depletion force is due to the expulsion of material.

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